



Inverse and hybrid approaches for disordered solids

DAVID DRABOLD, PARTHA BISWAS, D. K.
LIMBU, R. ATTA-FYNN, D. TAFEN, ANUP
PANDEY, BISHAL BHATTARAI, RAJENDRA
THAPA

Logic...

Modeling as a problem of inference



E. T. Jaynes
1922-1998

- 1) Conventional approach: Do your best simulation (big cell, fancy interactions, long simulation times, etc). *This is very different from the way the material was really made!* Compare to experiment, hope for the best. Write the paper.
- 2) There may be other information that would improve the realism of the simulation -- experimental, chemical or other information not included. *Why not use it in making the model?!*

Modeling paradigms and imposing *a priori* information

- 1) **Simulation:** Implement your best calculation (big cell, fancy interactions, long time evolution, etc). *Hope* that the results look like experimental ones.
- 2) **Information:** Try to invert the experimental data.
- 3) **Merge the two:** carry out simulation but impose the *a priori* (possibly experimental) information as part of the simulation.

Reverse Monte Carlo

Kaplow, McGreevy *et al.*

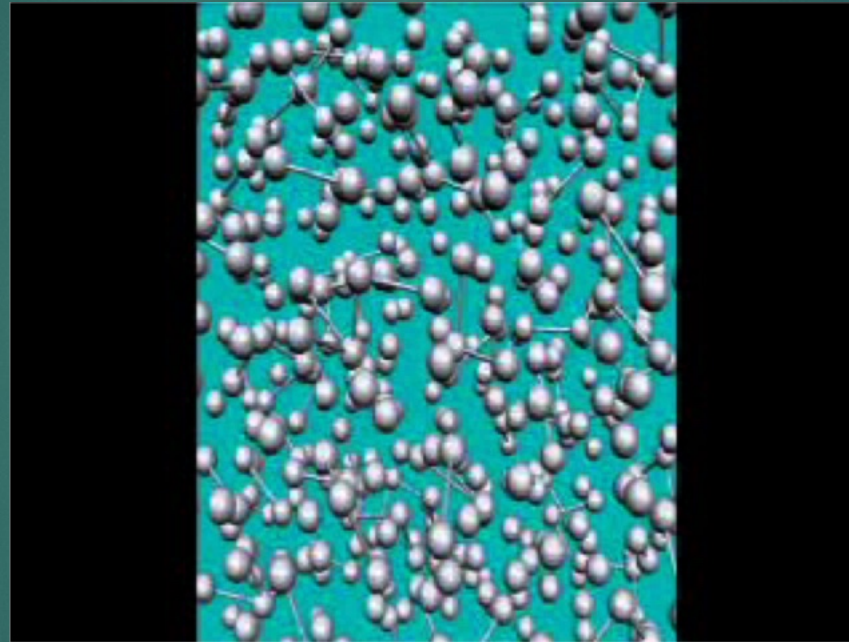
- ▶ **Information paradigm.** *What does experiment imply about the structure?*
- ▶ Reverse Monte Carlo~: put atoms in a supercell, move at random with Monte Carlo, keep moves if closer to experiment, accept with Metropolis probability if worse.
- ▶ Result: matches experiment by construction, but diffraction data **alone** is **insufficient** to produce a chemically realistic model. *Still, it is a clever idea -- use the information you have!*

RMC: Discussion

- ▶ Promising *if* additional information (constraints) are employed.
- ▶ Flexible, enables inclusion of *a priori* information.
- ▶ Constraints are tricky: we are imposing information, but we are potentially imposing errors – *the model is only as good as the information employed!*

RMC: order from chaos

RMC on α -Si (diffraction plus bond angle constraint)

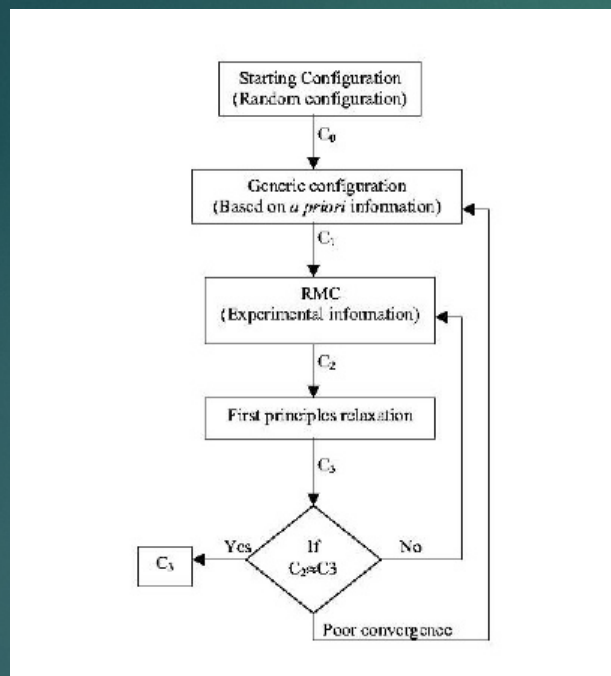


Credit: **Partha Biswas (about 2005)**

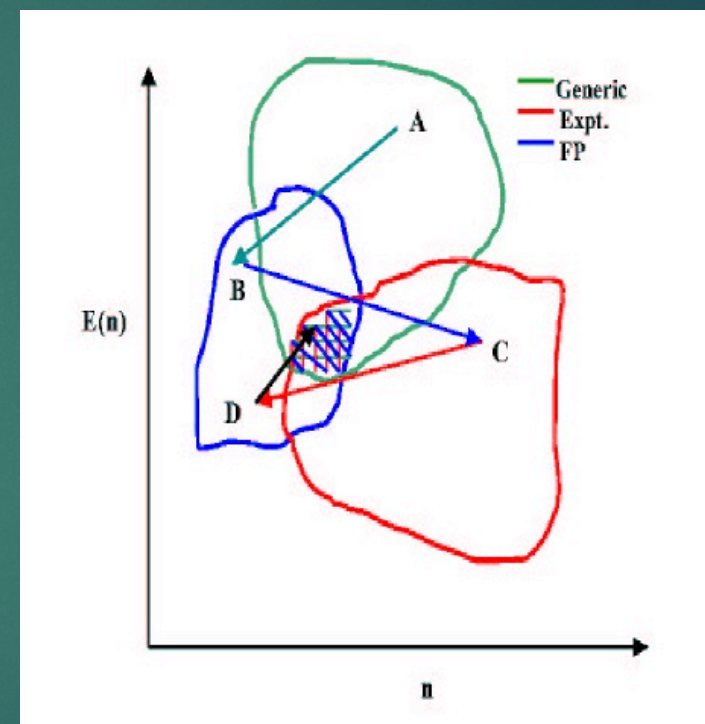
Experimentally Constrained Molecular Relaxation (ECMR)

- ▶ We want to include experimental data in MD: **merge RMC and molecular dynamics modeling.**
- ▶ Start from "experimentally realistic" subspace, self-consistently iterate between RMC and first principles relaxation.

ECMR: *implementation*



Flow chart for ECMR



Cartoon suggesting
ECMR convergence

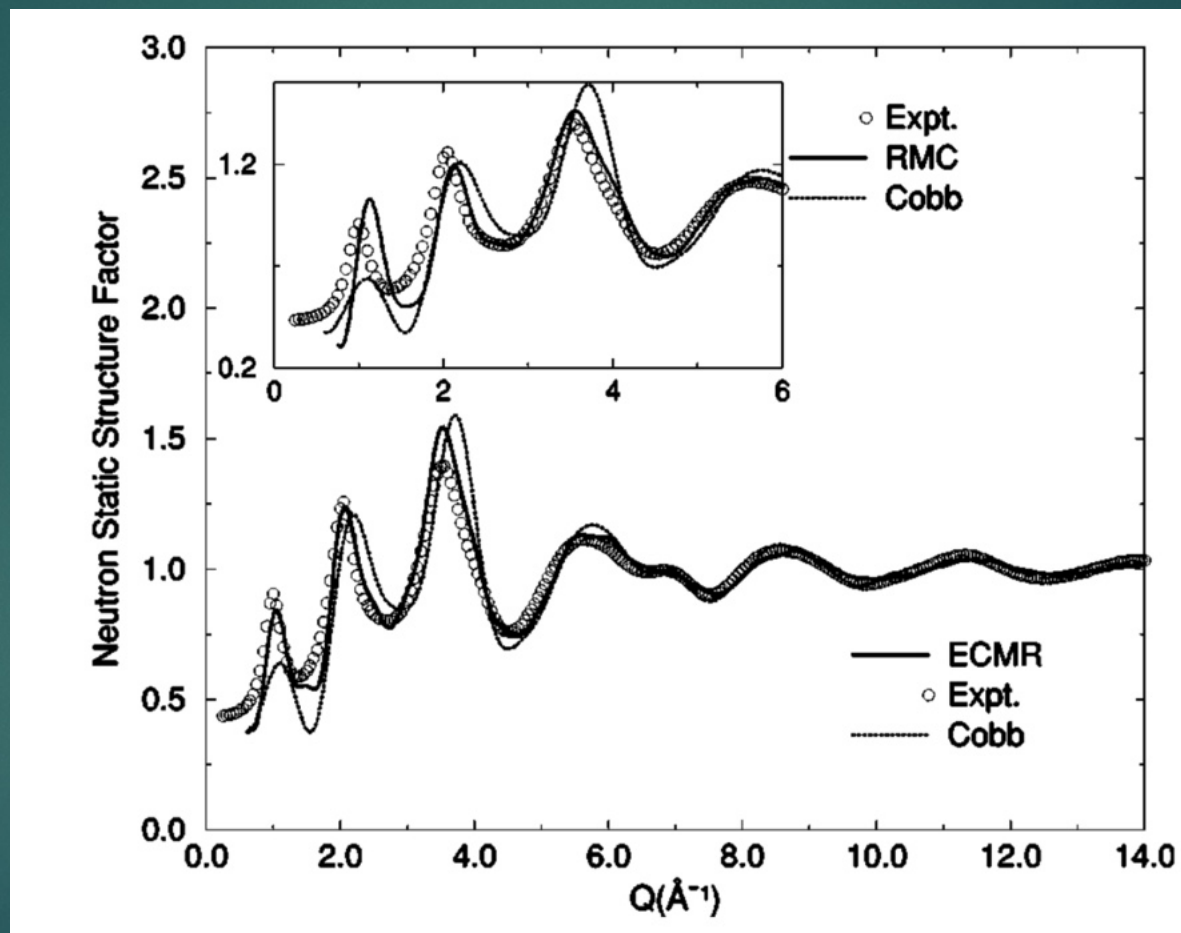
ECMR & GeSe₂

- ▶ GeSe₂: Classic chalcogenide glass, hard to model well, especially first sharp diffraction peak.
- ▶ Lets try ECMR:

Experimental input: Petri and Salmon partial structure factors.

Hamiltonian: Density functional, minimal basis, supercell with 648 atoms in unit cell.

ECMR: g-GeSe₂ results



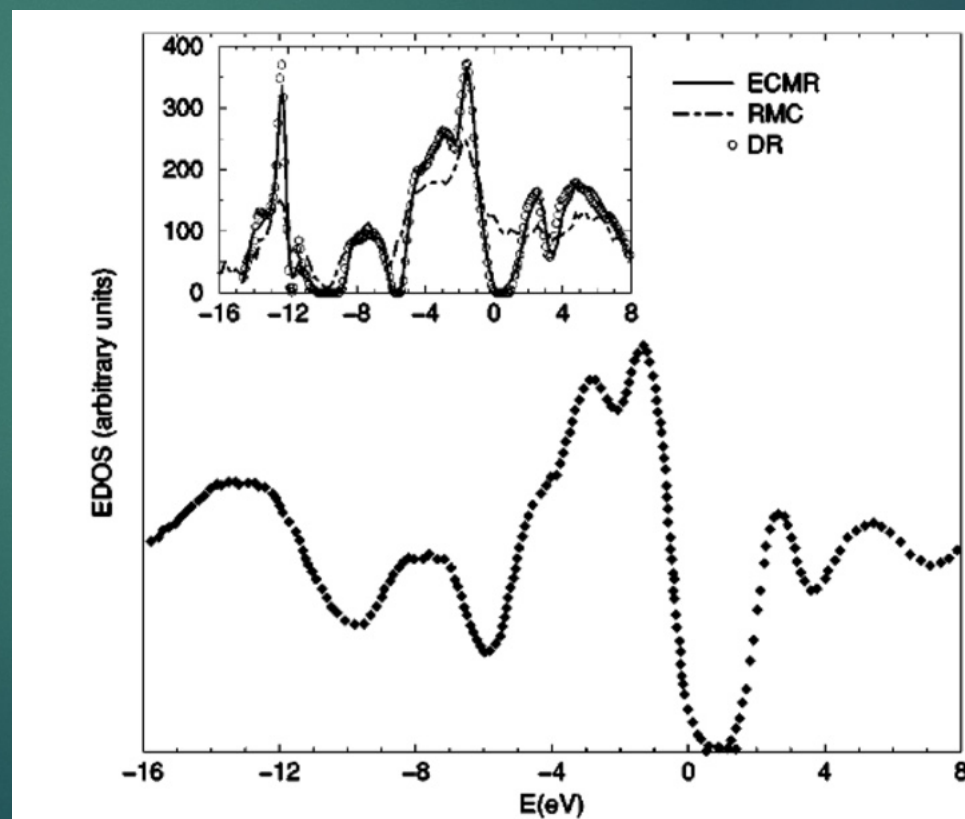
Static structure factor

Electronic structure: density of states

Note: a valid structural model **must** have a realistic electronic DOS.



Expt: XPS - *Bergignat et al*
PRB (1988), IPES- *Hosokawa*
et al, JPCM (1994)



ECMR: improving the convergence

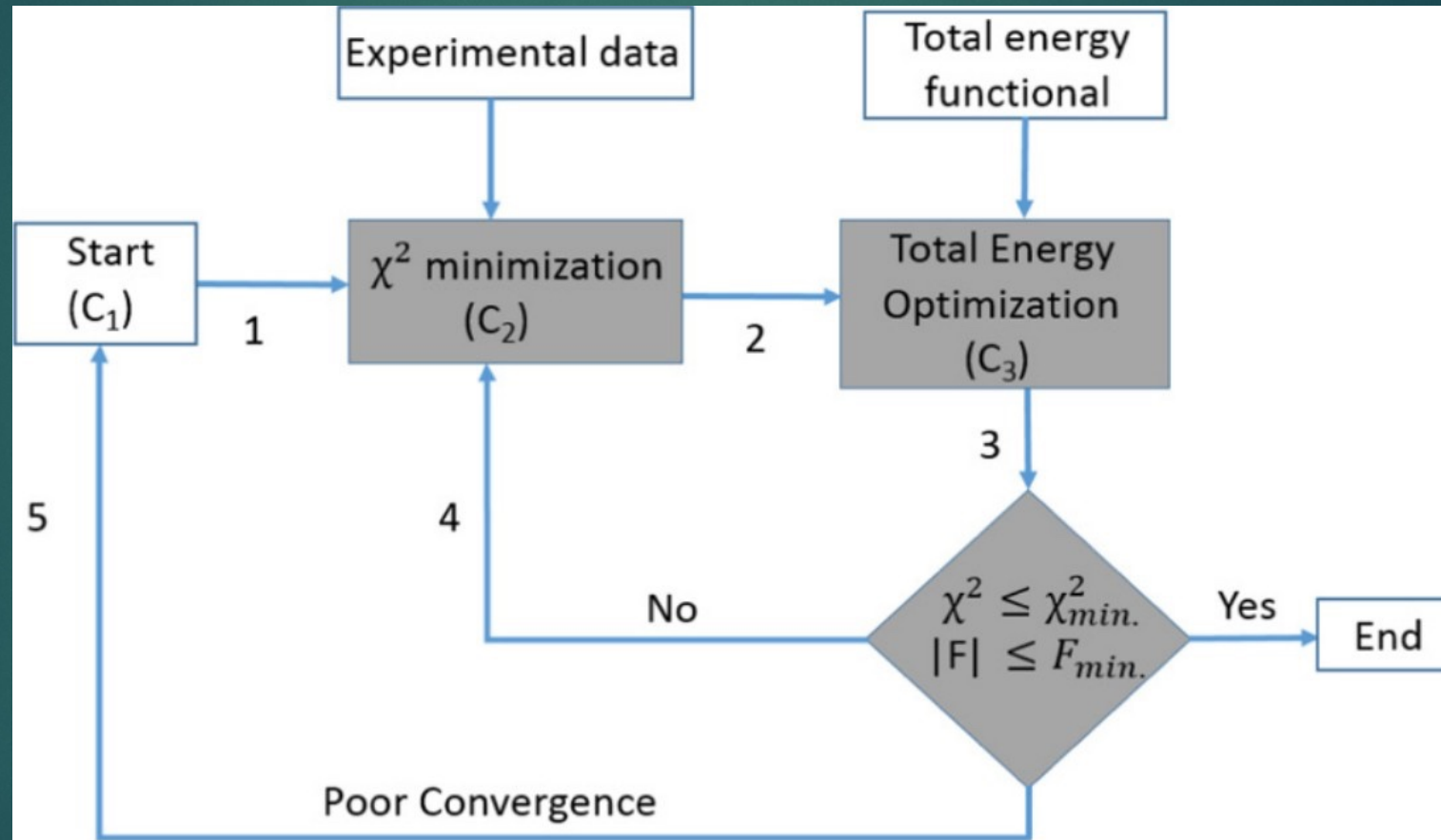
- ▶ ECMR is great – *if it converges!*
- ▶ Rather than doing a sequence of full relaxations, perform partial inversions, followed by partial relaxations.
- ▶ Iterate THAT!
- ▶ Seems to be more robust than ECMR.

Beyond ECMR: Force Enhanced Atomic Refinement (FEAR)

- ▶ Start with random model (assume density is known)
- ▶ Repeat to these two steps convergence:
 - Obtain N accepted moves from RMC [drives model toward experiment]
 - Take M conjugate gradients steps with energy functional [enforce chemistry]

Typically $N \sim 100$, $M \sim 1-5$. Always $N \gg M$.

Force Enhance Atomic Refinement (FEAR)



Partial Structural
minimization

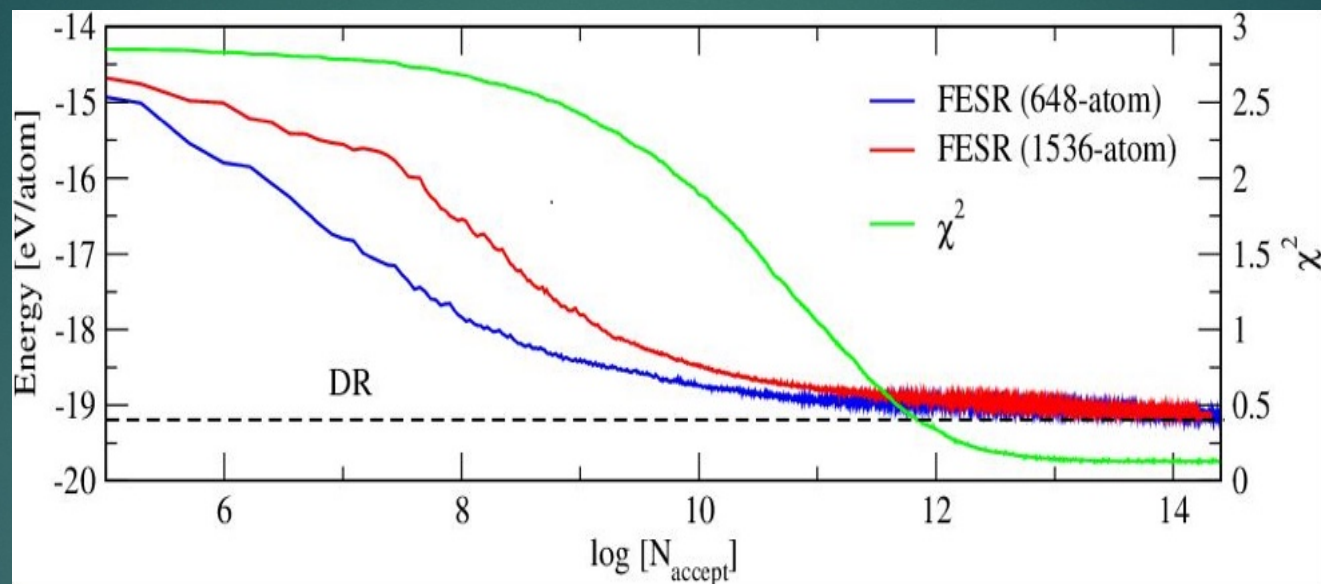
Partial Energy
minimization

Pandey et. al, Phys.RevB 94, 235208 (2016)

Example: FEAR for amorphous SiO₂

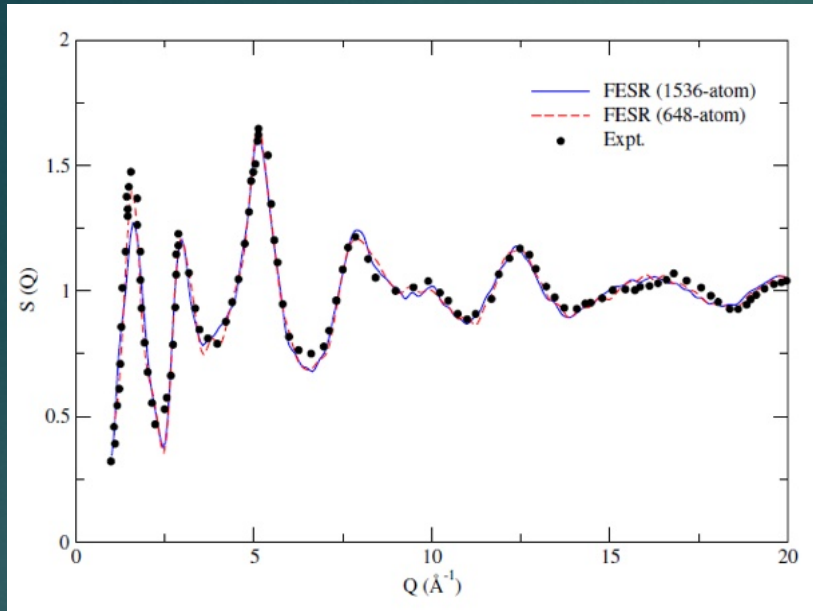
- ▶ Adopt 648-atom, 1536-atom models.
- ▶ Use the van Beest (BKS) potential (PRL, 1990). Start with **random** coordinates.
- ▶ After 100 successful RMC moves, move all the atoms along van Beest gradient – only one step, *not a full minimization*.
- ▶ *Repeat previous until convergence (fit and force) is achieved.*
- ▶ Need about 30,000 force calls

FEAR of silica

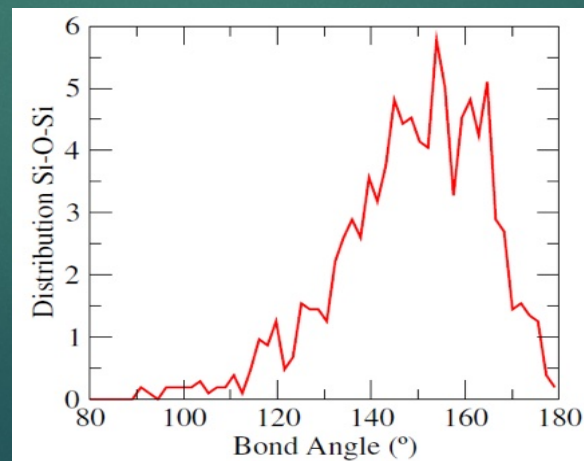
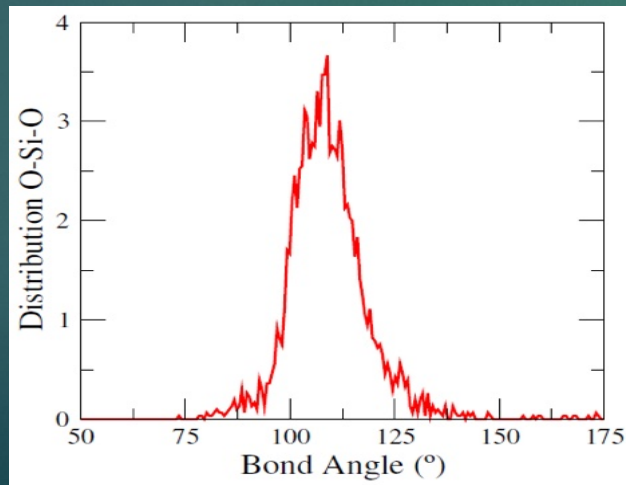


FEAR: minimization of error vs. experiment and total energy.

Results: silica



| Peak position (Å) | | | |
|-------------------|------|------|-------------------|
| atom-atom | FESR | MD | Expt. |
| Si-Si | 3.15 | 3.10 | |
| Si-O | 1.62 | 1.62 | 1.610 ± 0.050 |
| O-O | 2.64 | 2.64 | 2.632 ± 0.089 |



| Bond Angle (°) | | | | |
|----------------|-----------------|---------------|-------------|--------------|
| | FESR | MD | Expt. | DR |
| O-Si-O | 109.5 (15.6) | 109.6 (10) | 109.5 | 109.5 (9) |
| Si-O-Si | 154.3 (27.8) | 142.0 (25) | 144 (38) | 140 (25) |

Ab initio FEAR – use DFT (VASP or SIESTA) as energy functional

Key to making a general method – to structurally invert a very wide range of materials from diffraction – need general accurate interatomic interactions to unbiased provide chemical information.

First example: silicon and SIESTA

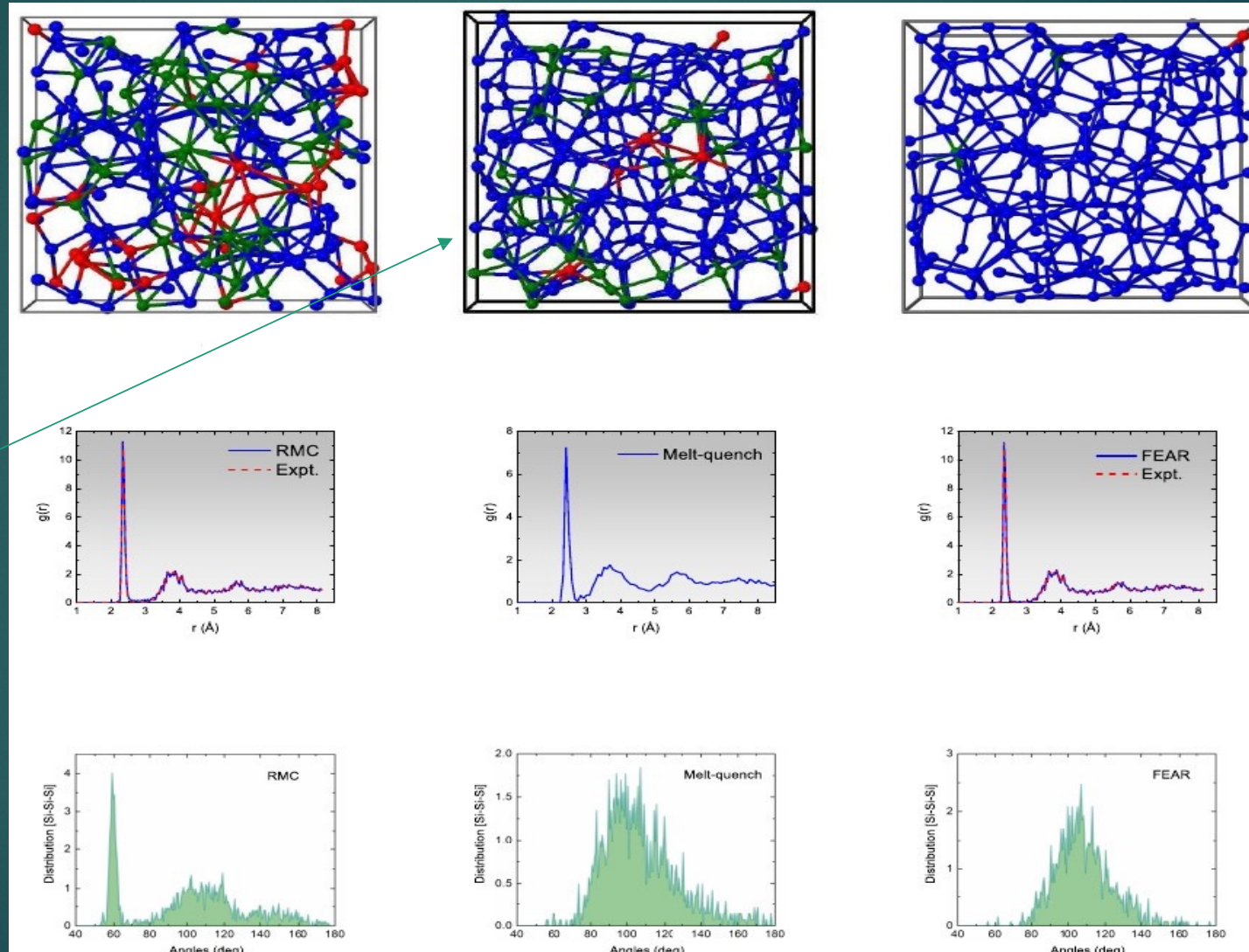
RMC Melt Quench FEAR

RMC

Melt-Quench

Blue: 4-fold
Green, Red are
coordination
defects

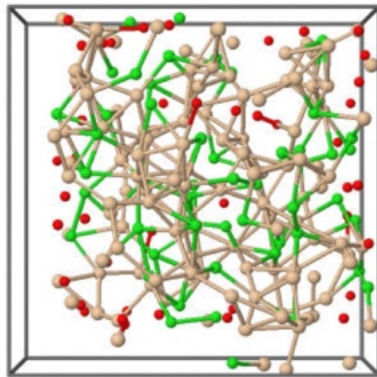
FEAR



FEAR: α -Si animation and details

Force-enhanced Atomic Refinement:

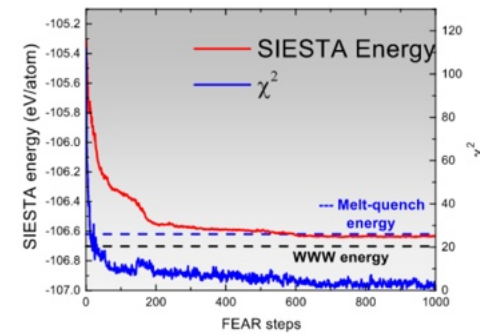
Evolution of 216-atom model amorphous Si starting from random initial configuration with beige sphere representing (correctly) four-fold atoms, green over-coordinated and red under-coordinated.



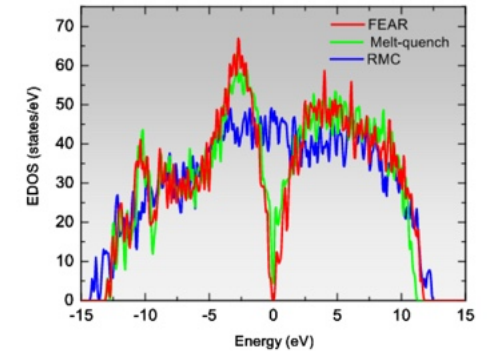
RED : Si (<4)

GREEN : Si (>4)

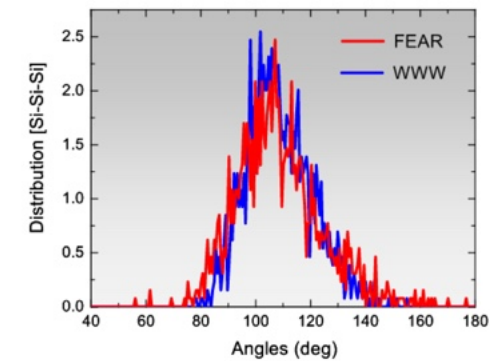
BEIGE : Si (=4)



a)



b)



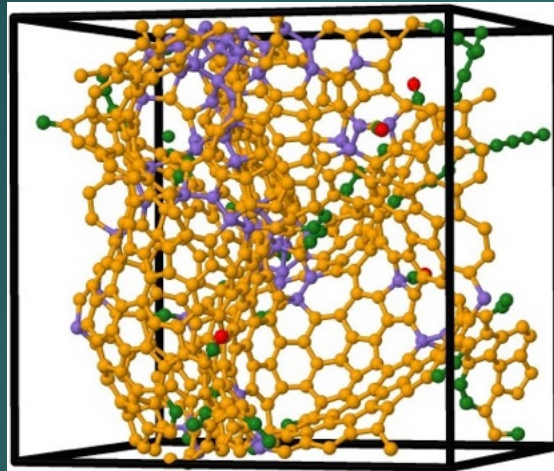
c)

Example: Amorphous carbon across densities

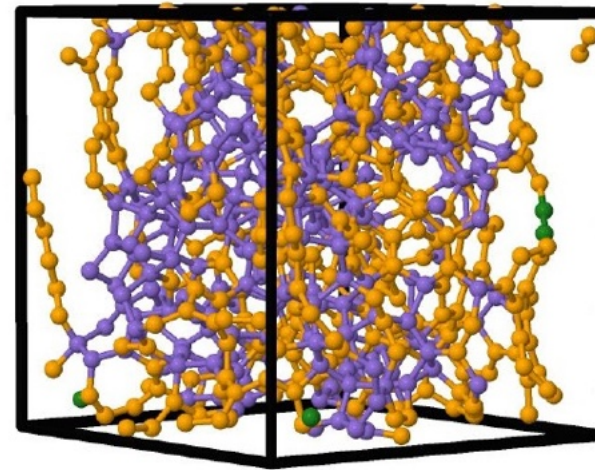
- ▶ Hard: Carbon happily sp^3 , sp^2 or even sp bonds. Need a good potential.
- ▶ Wealth of experiments to check against.
- ▶ We carry this out with largish models (up to 800 atoms), SIESTA as energy functional. Then relax final models with VASP (little change).

Amorphous Carbon across densities

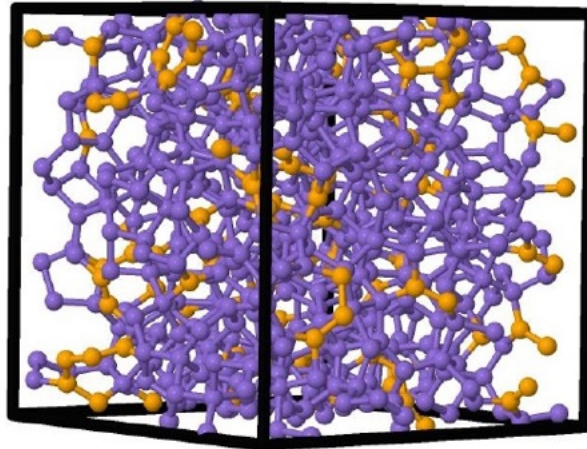
648 atoms



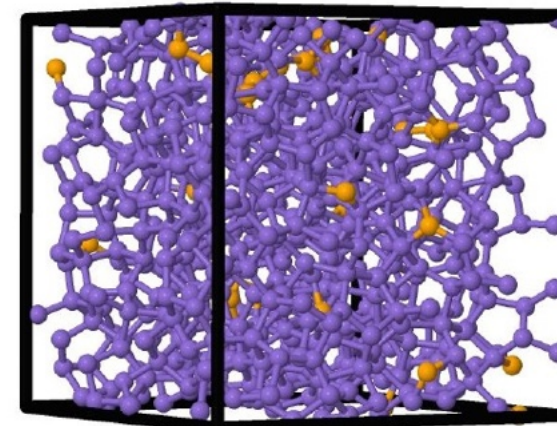
$\rho = 0.95 \text{ g/cm}^3$



$\rho = 2.44 \text{ g/cm}^3$



$\rho = 2.99 \text{ g/cm}^3$

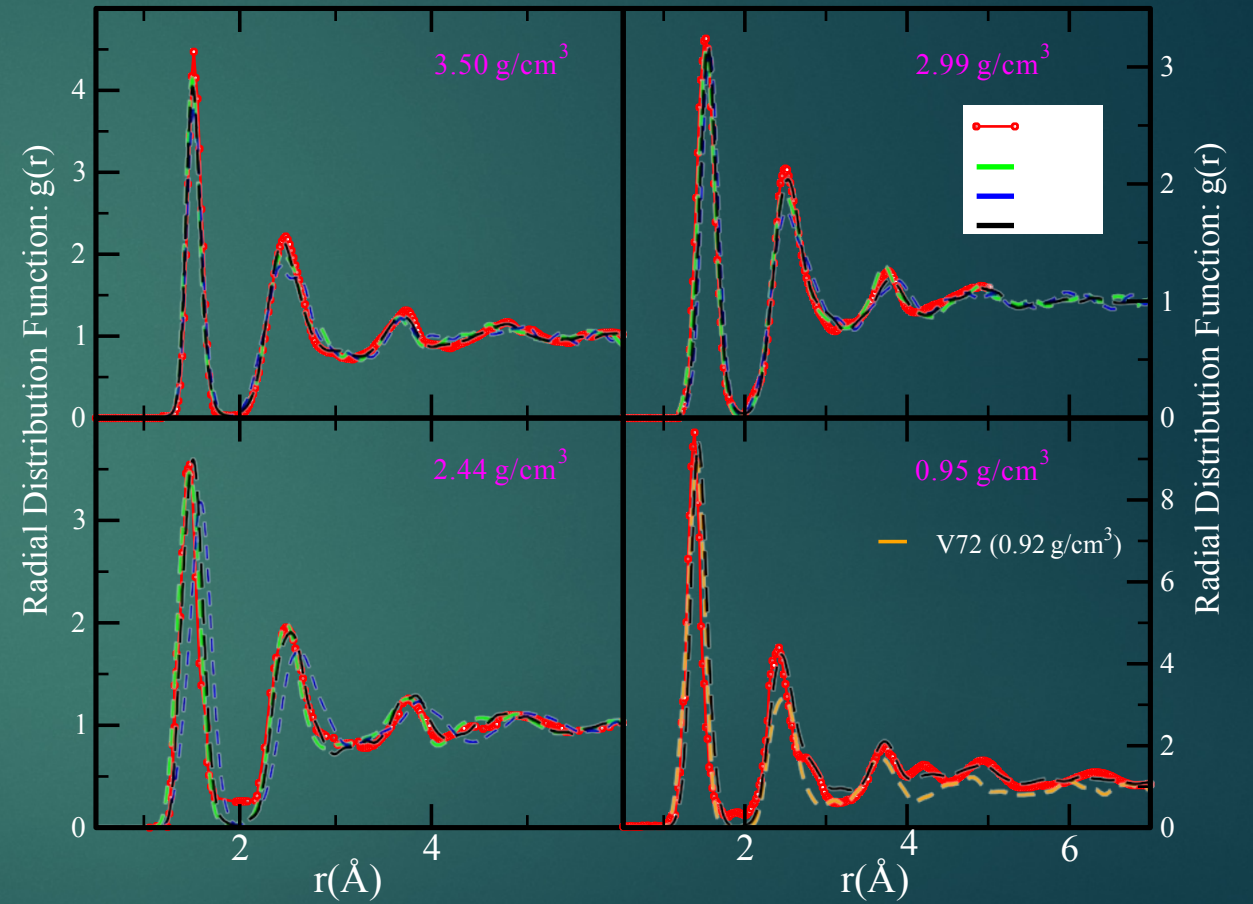
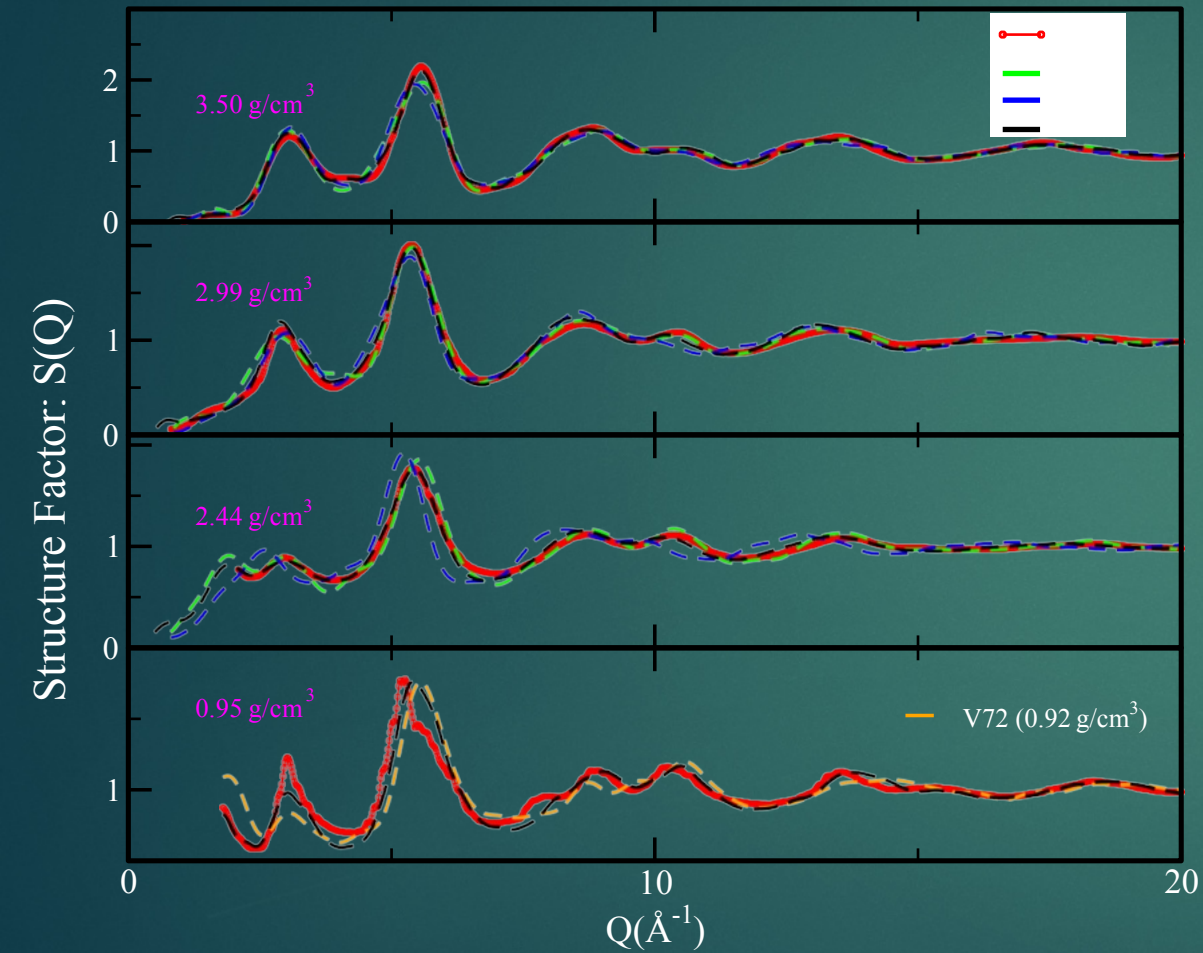


$\rho = 3.50 \text{ g/cm}^3$

Purple
(sp³),
Orange
(sp²),
Green
(sp)

Bhattacharj, Pandey & DAD, Carbon, 131 168 (2018)

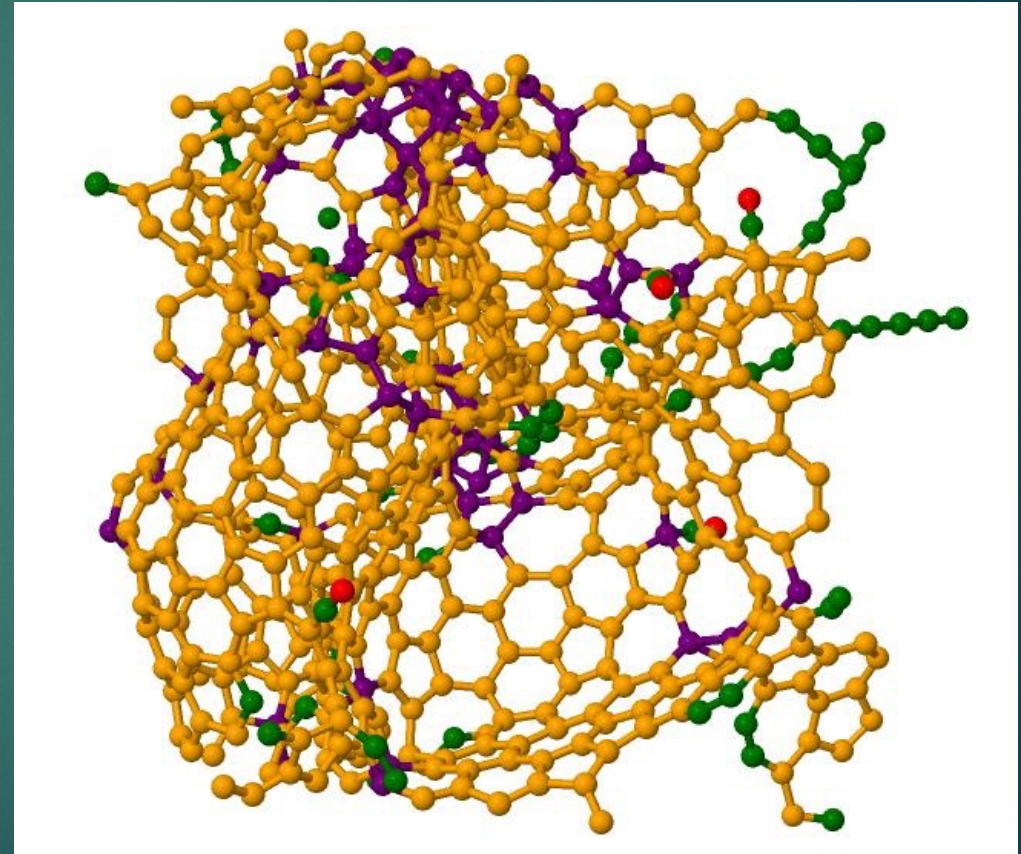
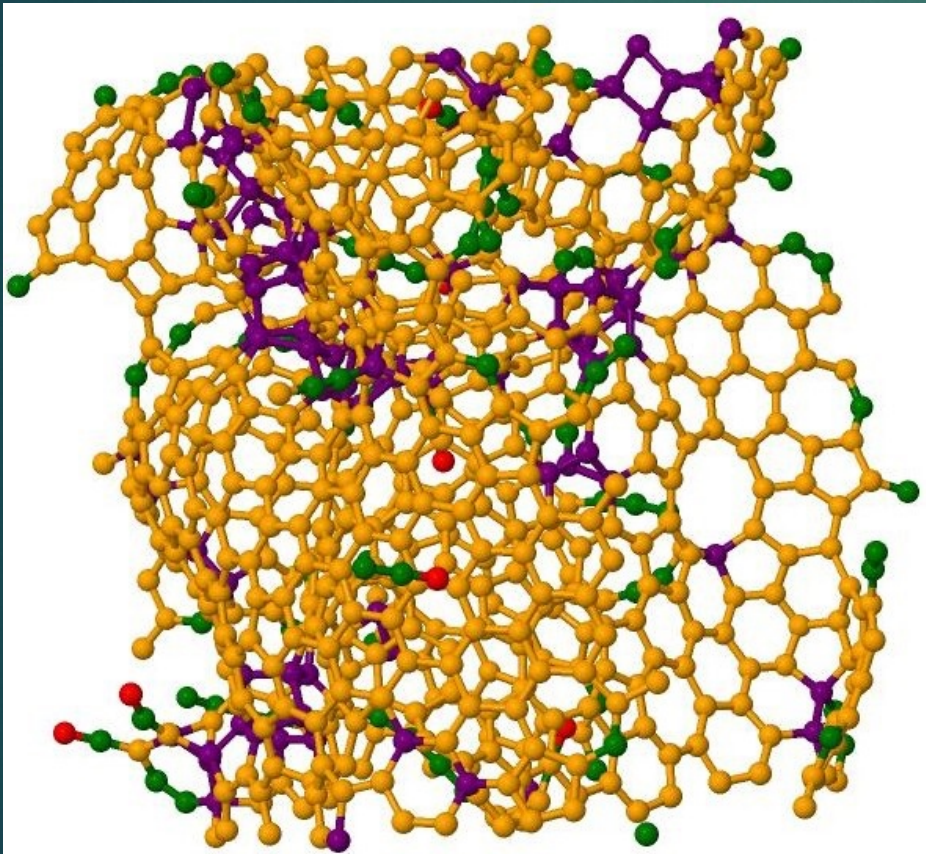
Amorphous Carbon



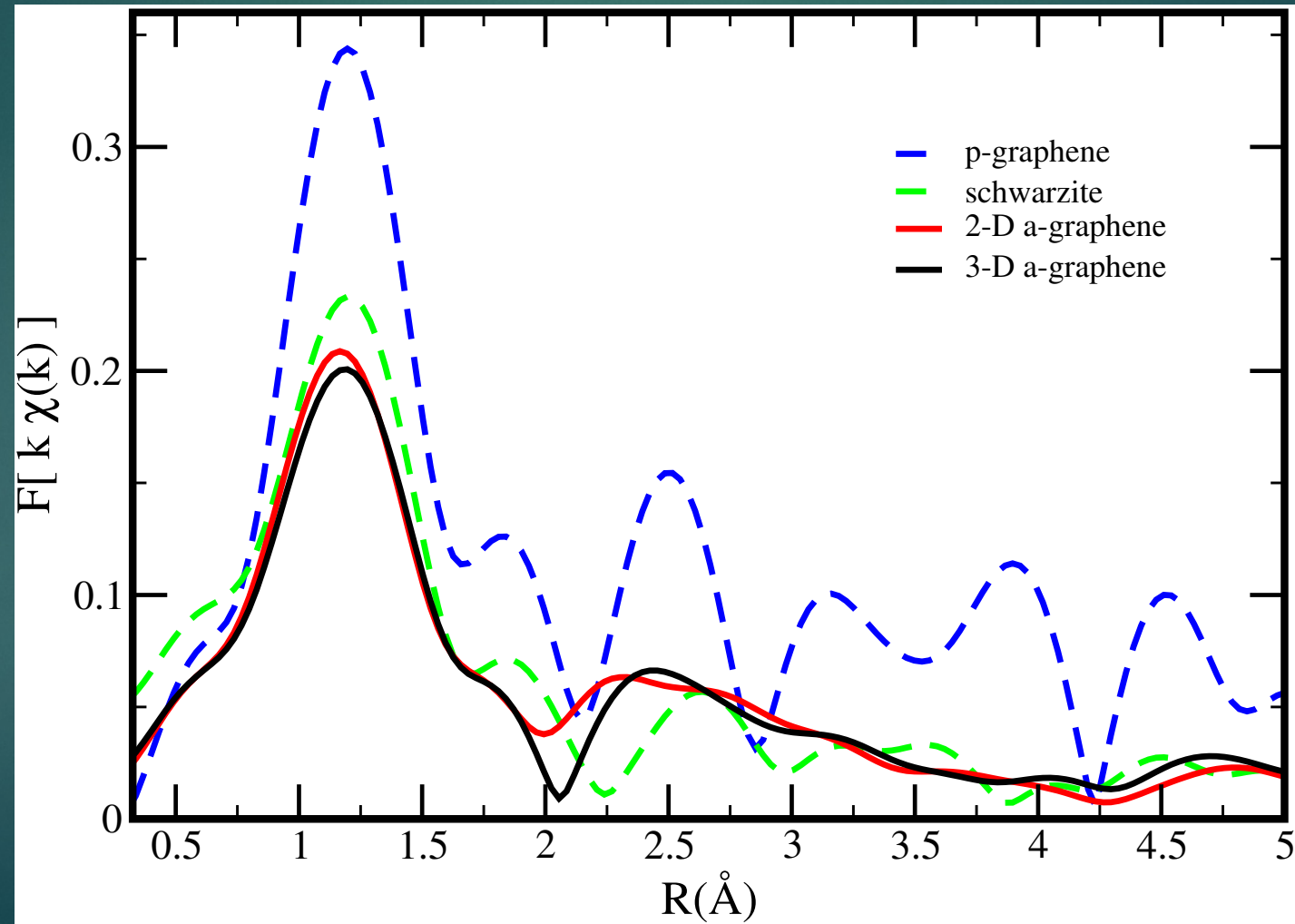
Bhattarai et. al, Carbon (2018)

Low density (0.95 gm/cc) FEAR Carbon (800-, 648-atom models)

Purple (sp^3), Orange (sp^2), Green (sp)



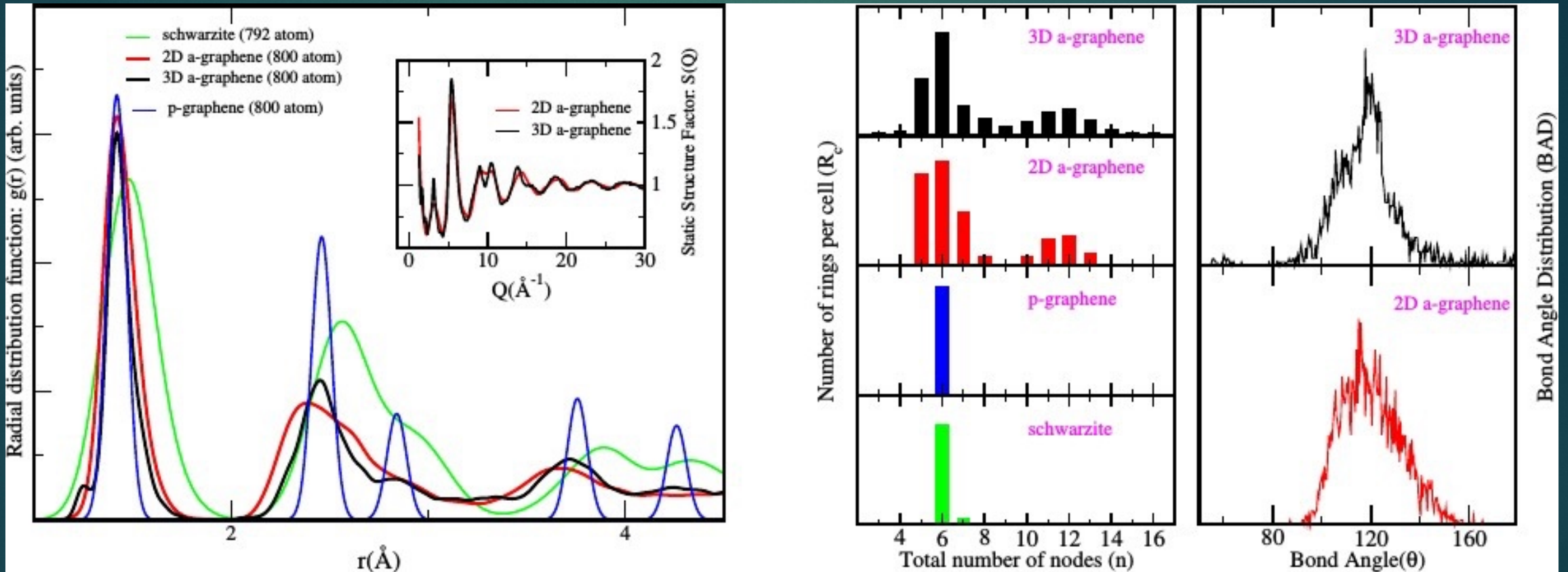
A prediction: EXAFS of 0.95gm/cc
 α -C. Fairly small differences...



Comment

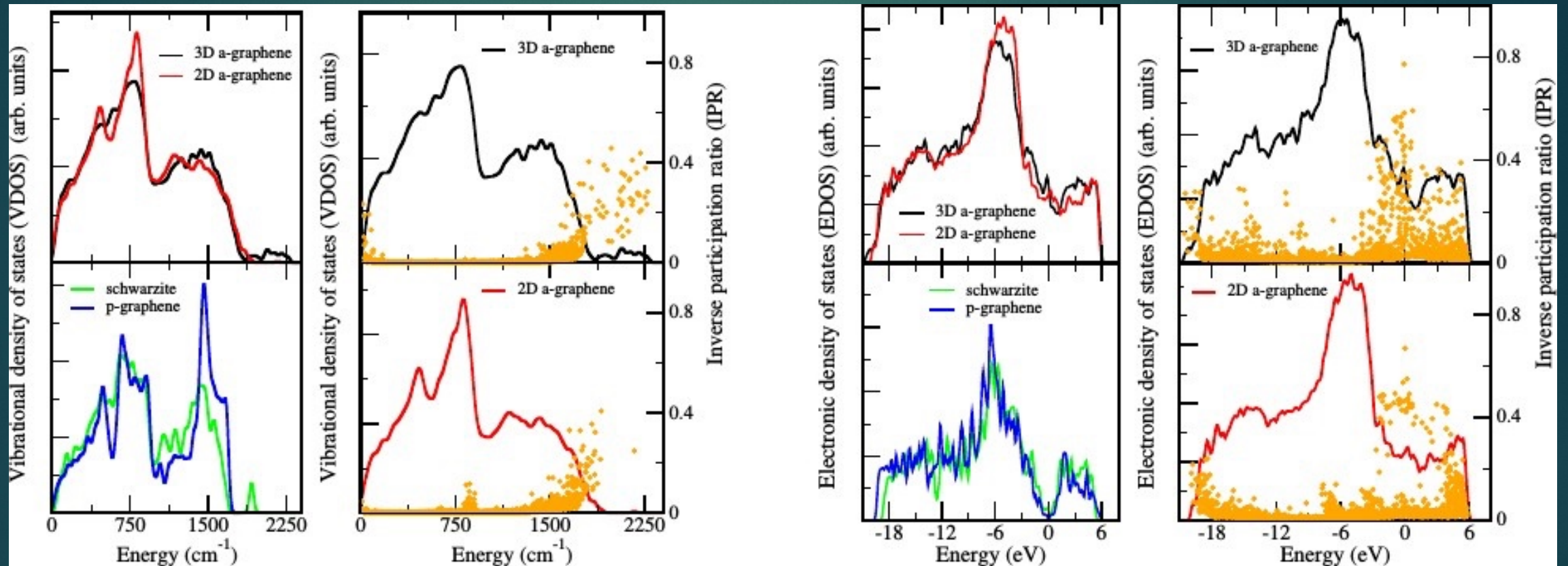
- ▶ Amorphous C with density near 1 gm/cc is a form of three-dimensional graphene: warped, wrapped sp^2 sheets including ring disorder (pentagons, hexagons, heptagons) and also with sp and sp^3 defects.

Structural Comparison

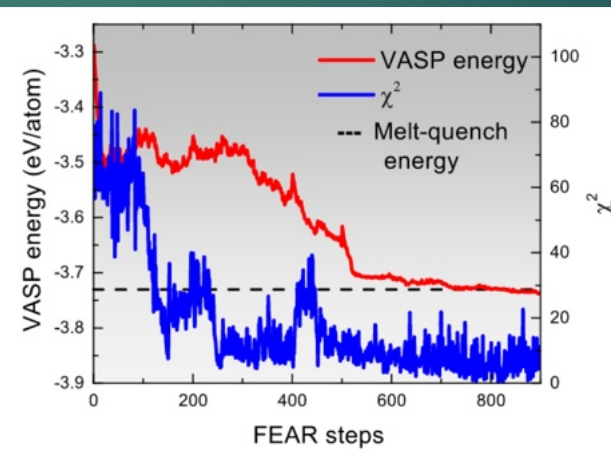
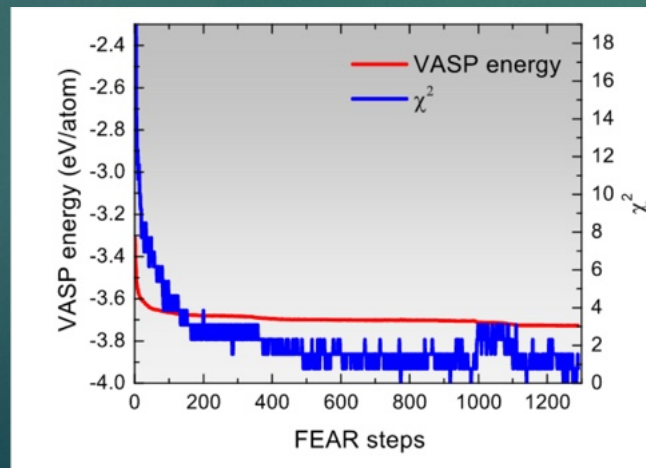
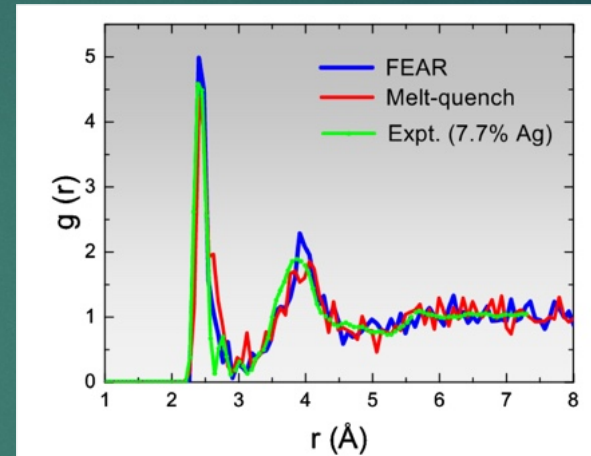
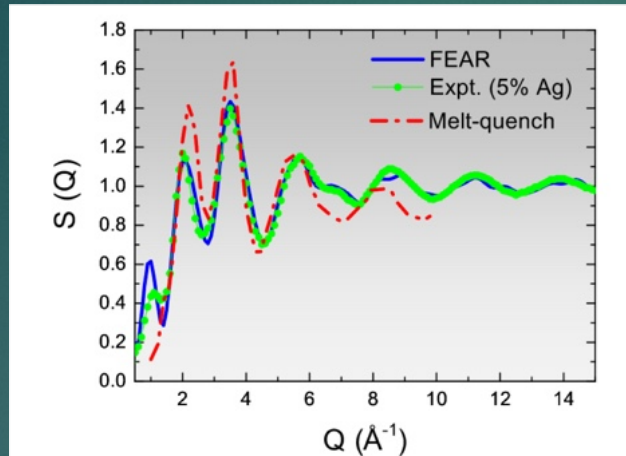


Bhattarai et. al, PRL submitted (2018)

Electronic and Vibrational Properties



FEAR: Ag-doped chalcogenides, $[(\text{GeSe}_3)_{1-x}\text{Ag}_x]$ $x=0.05, 0.077$ data: Zeidler and Salmon (Bath) VASP, A. Pradel group (Montpellier)



Conclusion (FEAR)

- Efficient: Fewer calls to force code.
- Robust convergence: Really works [α -Si, α -C (0.95-3.5 gm/cc), GeSeAg materials]. We're trying a metallic glass, fiddling with EXAFS too – $\text{Pd}_{40}\text{Ni}_{40}\text{P}_{20}$ (nothing to report yet!). Used empirical pots, tight-binding, SIESTA and VASP. Routinely produces (slightly) lower total energies than a reasonable melt quench.
- Easy: if you know RMC and VASP, this is essentially a shell script.
- It is GENERAL.